



Fluctuations critiques dans un fluide ternaire s'opposant à la complexation dans l'extraction liquide- liquide de cations trivalents.



*Thomas Zemb, Asmae el Maangar , Jean-François Dufrêche &
Stéphane Pellet Rostaing (ICSM)*

Werner Kunz (U Regensburg)





Critical fluctuations in a ternary fluid opposing complexation in the liquid-liquid extraction of trivalent cations.

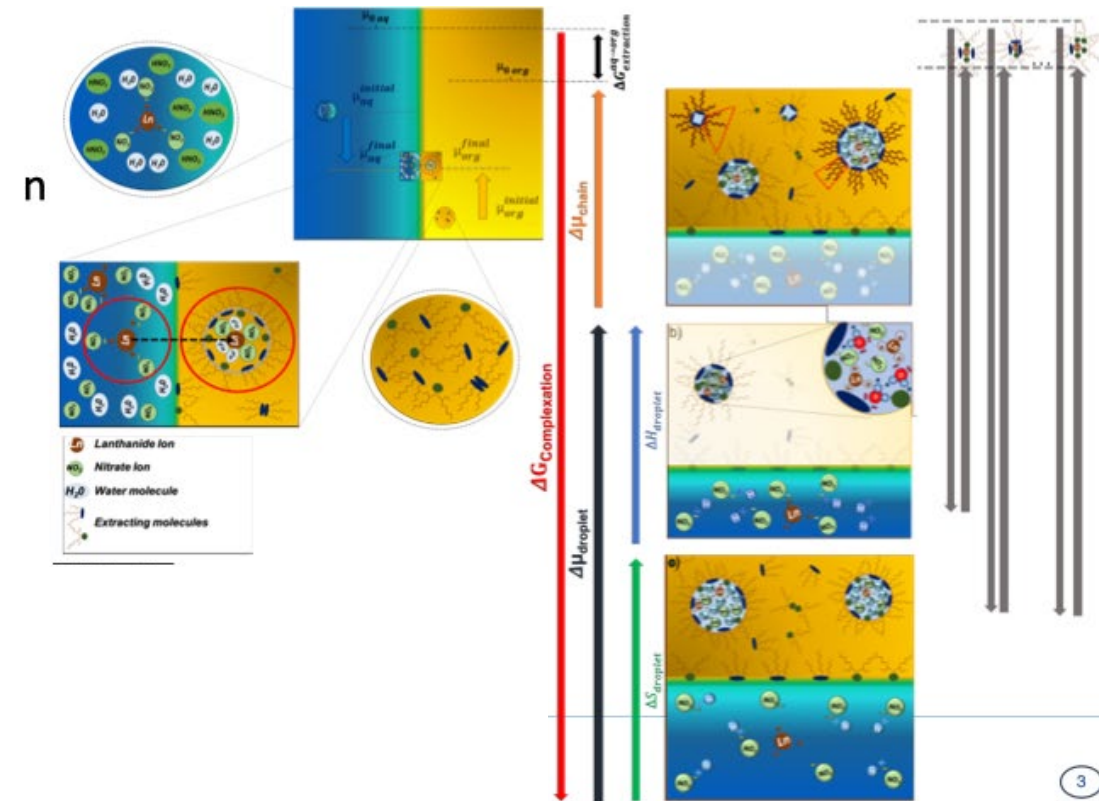
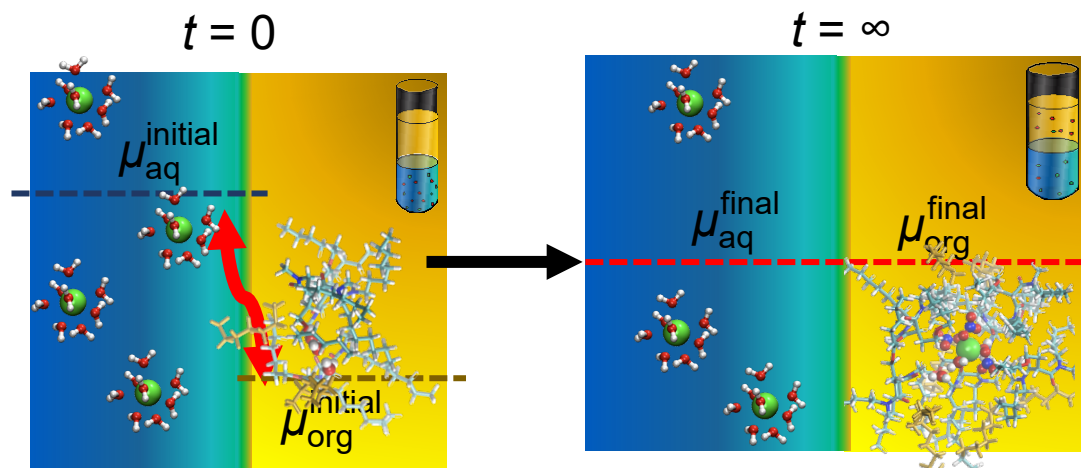


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Extraction/stripping view as a transfer between complex fluids



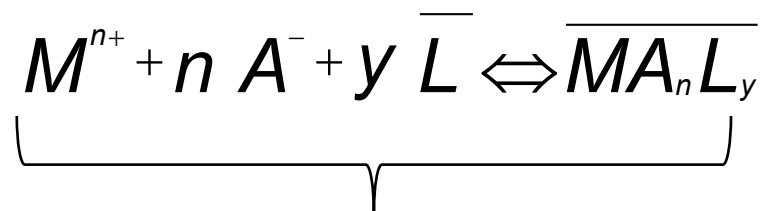
Recycling metals by controlled transfer of ionic species between complex fluids: en route to "iencas". « ThZ et al Colloid Polym Sci (2015) 293:1–22

Gibbs energy of the transfer : fundamental equation of ienaics



$$-RT \ln D = \Delta_{\text{tr}} G^{\circ} = \mu_{\alpha}^{\circ, \text{org}} - \mu_{\alpha}^{\circ, \text{aq}}$$

The approach last century (> 20.000 papers) : » supramolecular chemistry » & » supramolecular chemistry »



$$\Delta G_{\text{transfer}} = \Delta G^{\circ}_{\text{complex}} + \Delta G^{\circ}_{\text{solvent/complex mixing}}$$

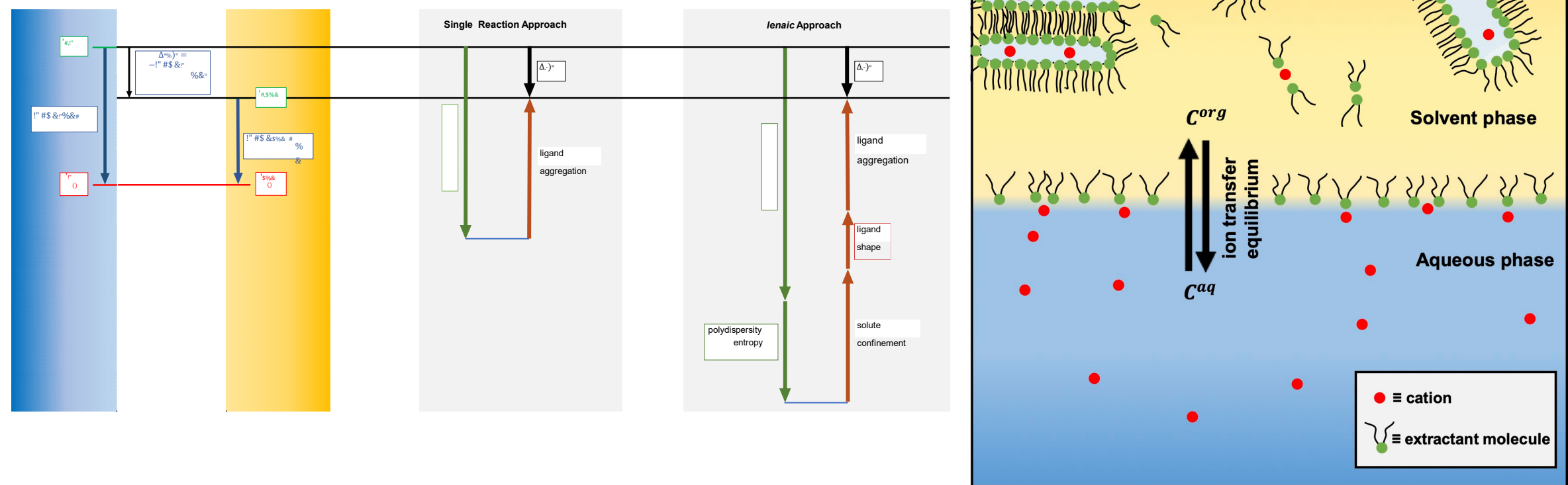
The « ienaics » approach (< 100 papers)) : take into account beyond « nearest neighbour enthalpy (entropy) i.e. colloidal chemistry

$$\Delta_{\text{tr}} G^{\circ} = \Delta_{\text{tr}} G^{\circ}_{\text{complex}} + \Delta_{\text{tr}} G^{\circ}_{\text{confin}} + \Delta_{\text{tr}} G^{\circ}_{\text{apolar}} + \Delta_{\text{tr}} G^{\circ}_{\text{lagg}} - T \Delta_{\text{tr}} S^{\circ}_{\text{pdisp}}$$

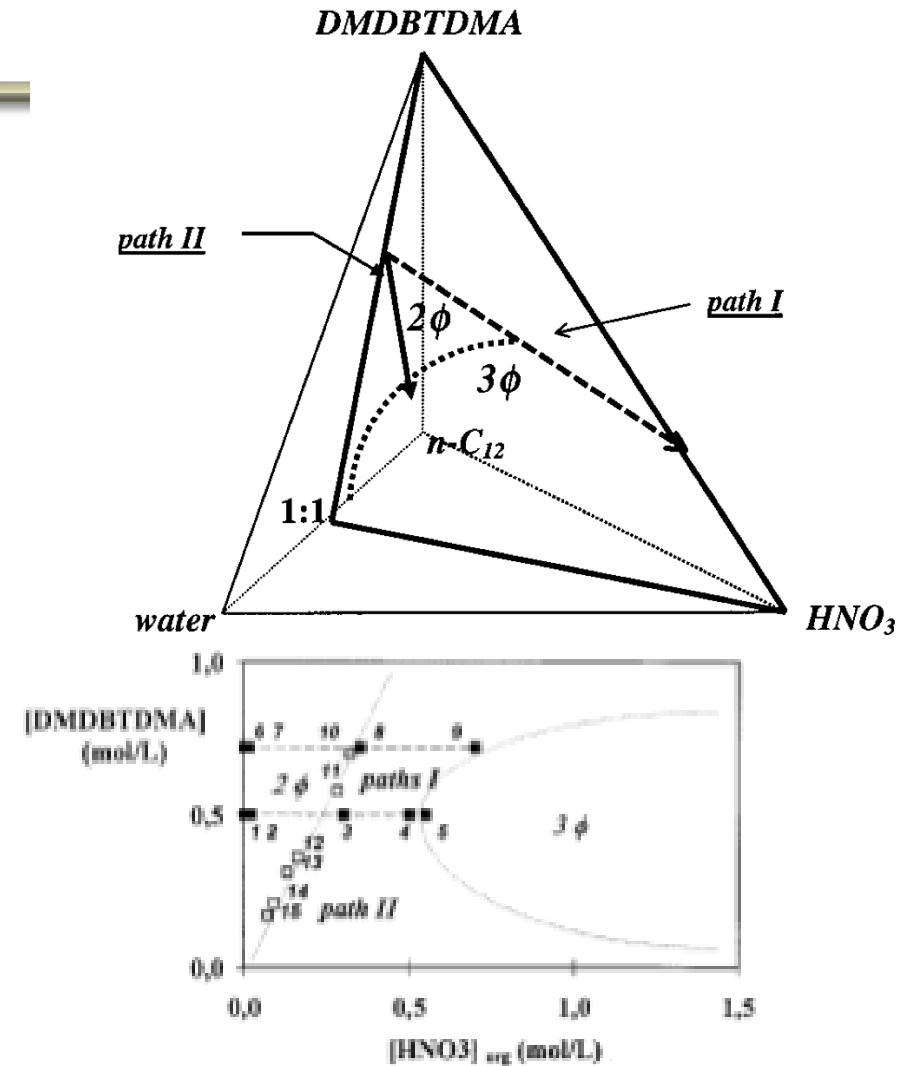
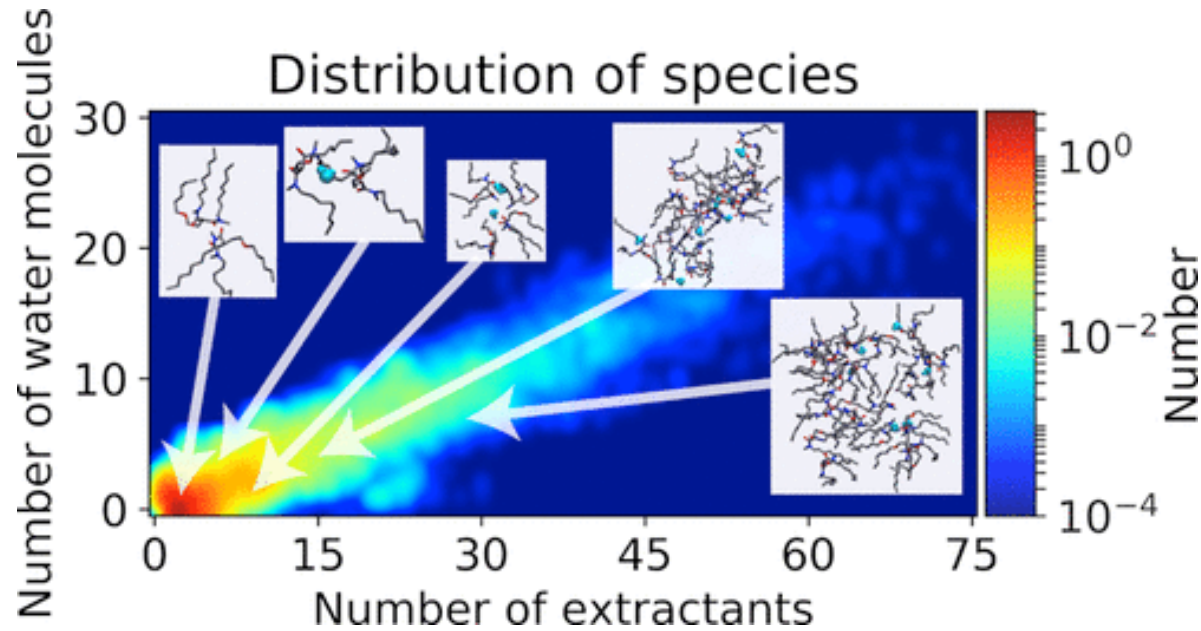
M. Spadina, JF Dufrêche et ThZ: « **Synergistic Solvent Extraction Is Driven by Entropy** »

ACS-nano (2019) 13, pp. 1375-13748

Extraction/stripping view as a transfer between complex fluids



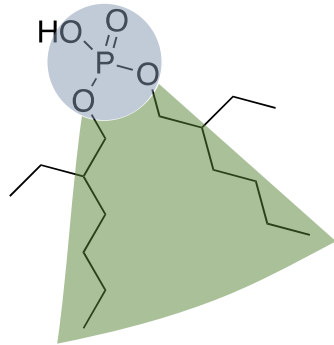
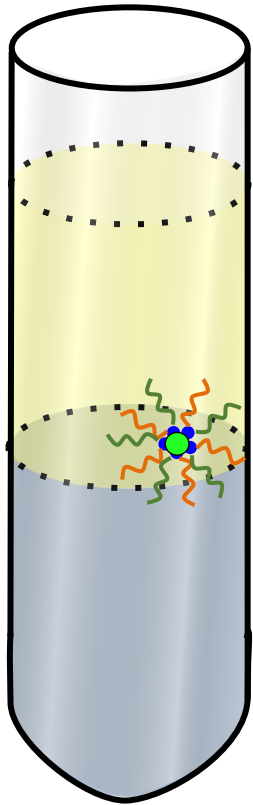
There are two general problems...



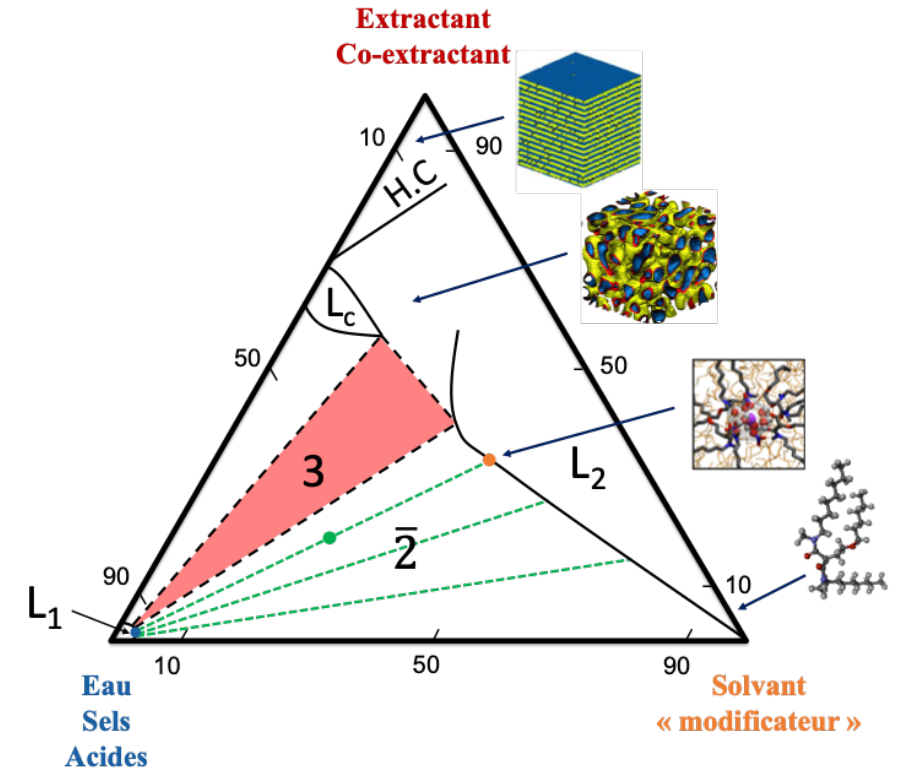
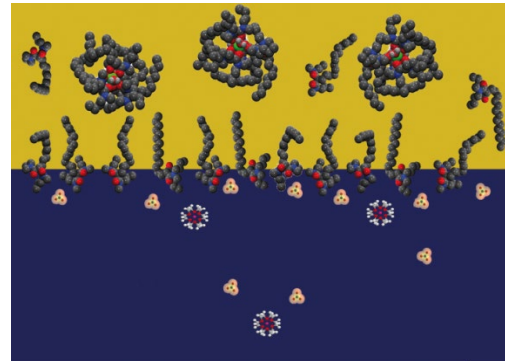
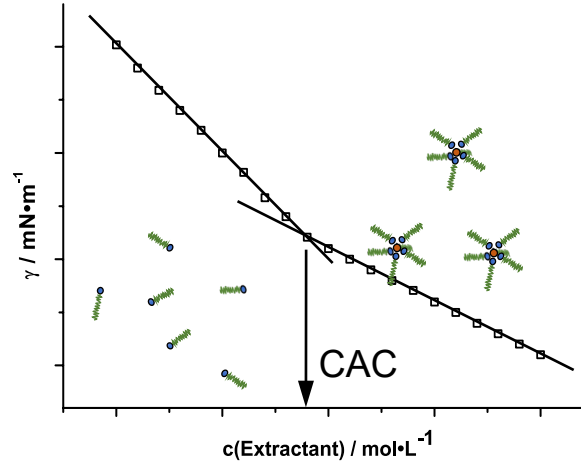
There is full distribution of aggregates $f(N_e, N_w)$ in the solvent; with different extracting powers the yield and occurrence of spontaneous emulsification is influenced by proximity of critical points

Recycling metals by controlled transfer of ionic species between complex fluids: en route to "iencs". « ThZ et al. Colloid Polym Sci (2015) 293:1–22

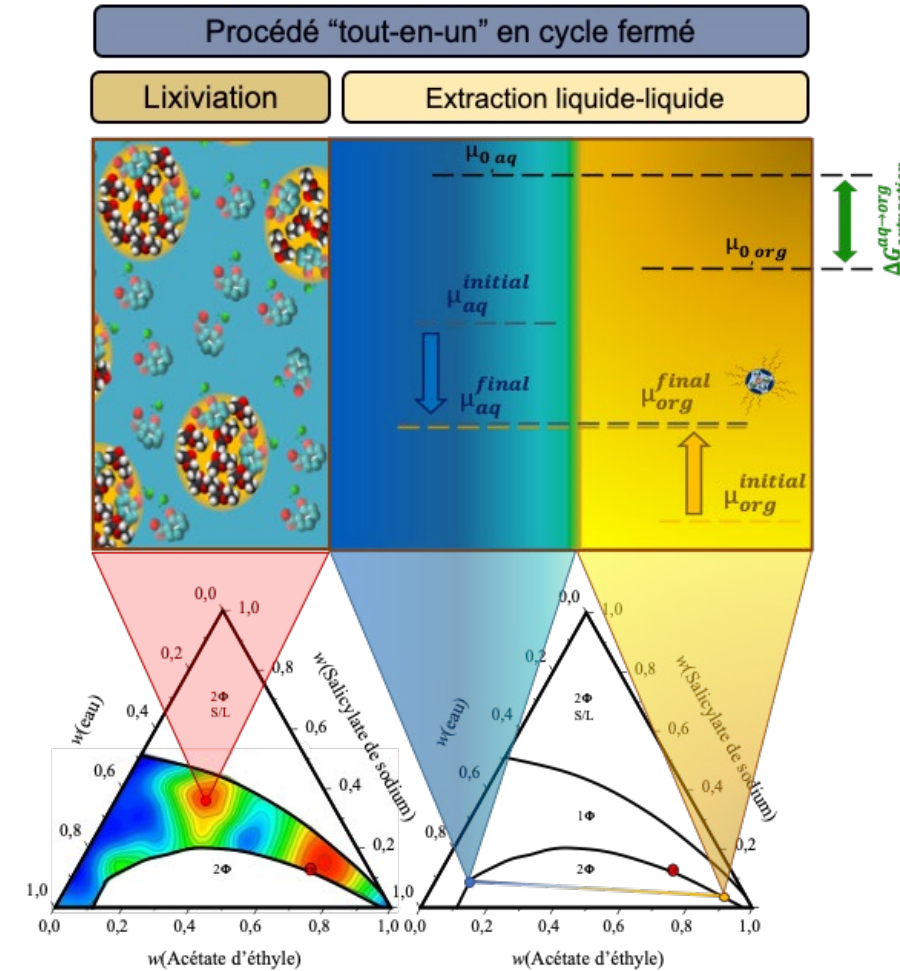
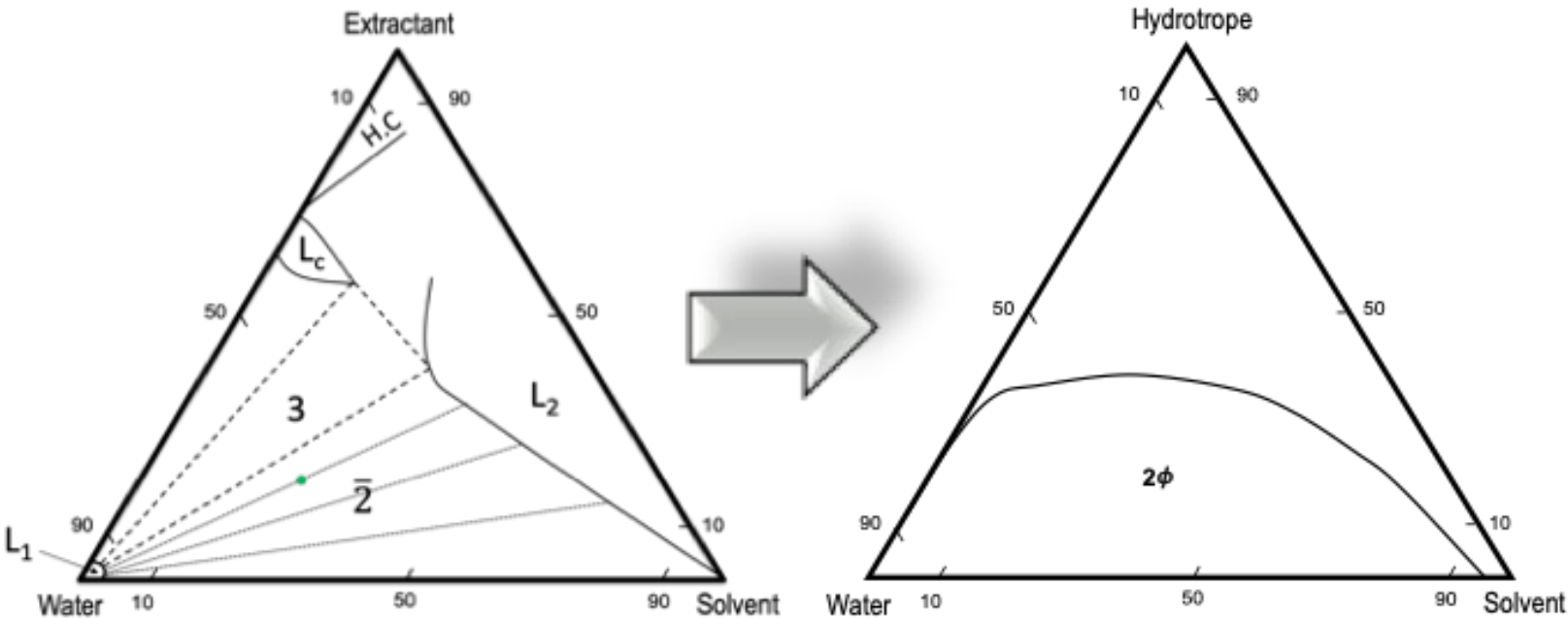
Classical extraction: water- solvent (extractant-diluent-modifier)



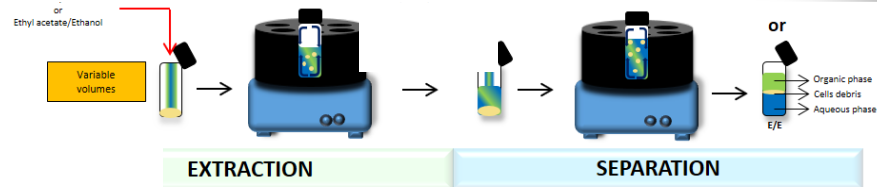
$$p_0 = \frac{v_{apolar}}{a_o \cdot l_c}$$



Towards Hydrotrope assisted extraction ?



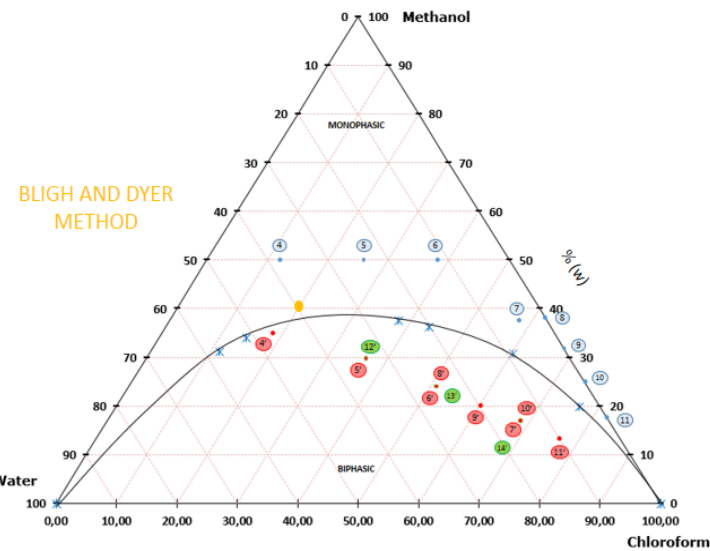
Using hydrotropes works well also for extraction: separation of lipids versus via a green Bligh&Dyer (Folch) method



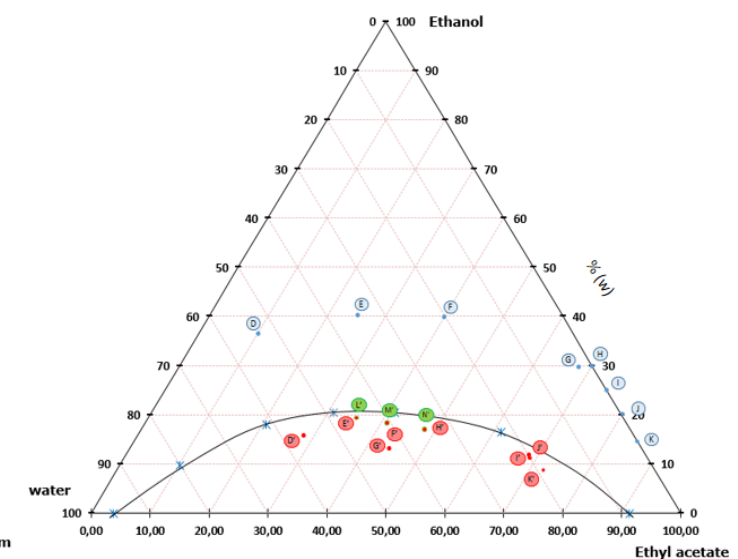
Aim is to extract all lipids (phospho- sphingo- glyco-) whatever their log p from glucids and proteins in cytoplasm of bacteria and other biomass

> 60.000 citations of « standard protocol » (no predictive theory before 2016)...

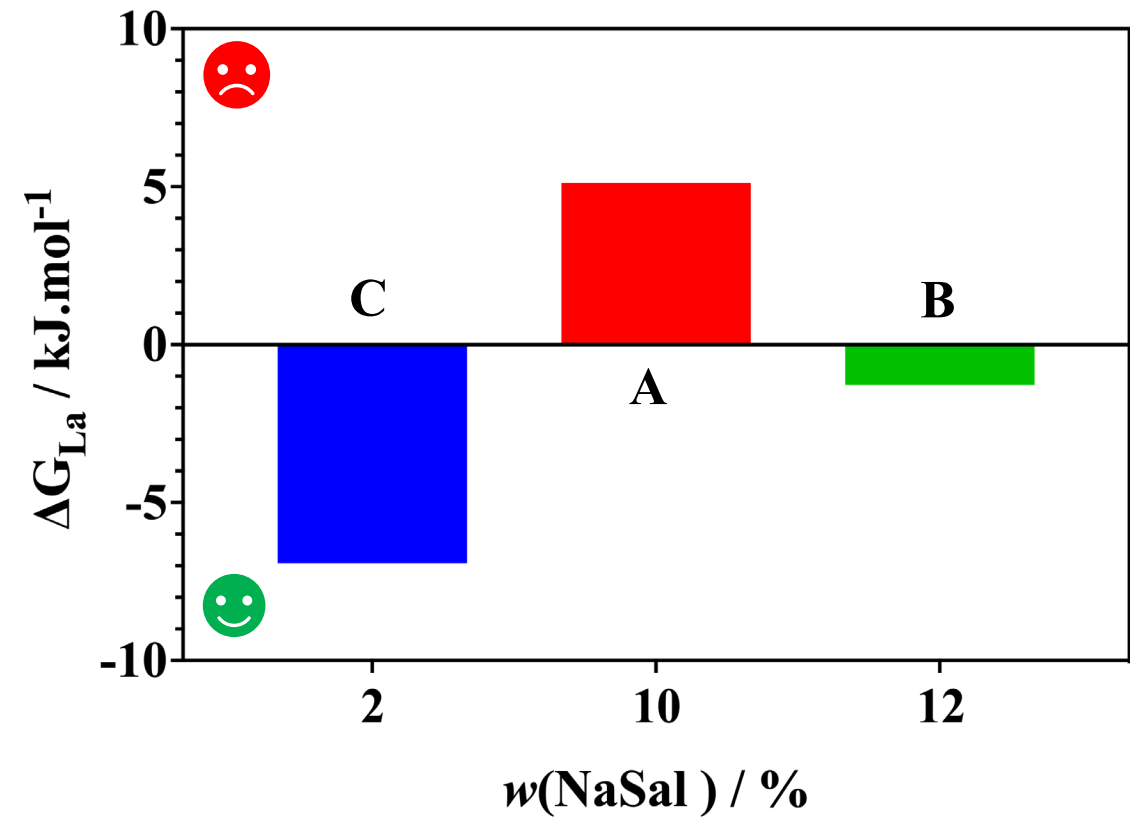
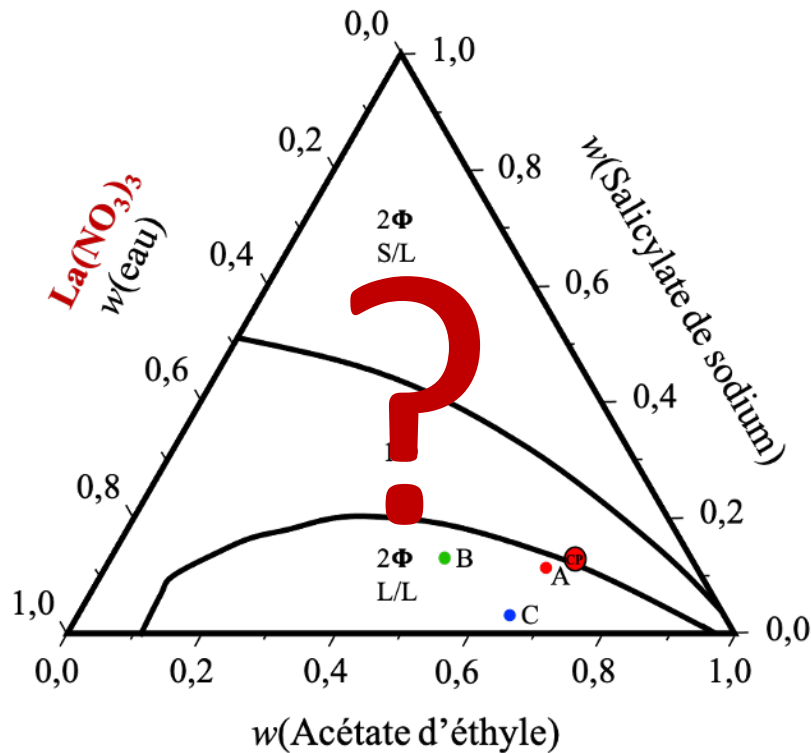
CLASSICAL BLIGH AND DYER



GREEN BLIGH AND DYER

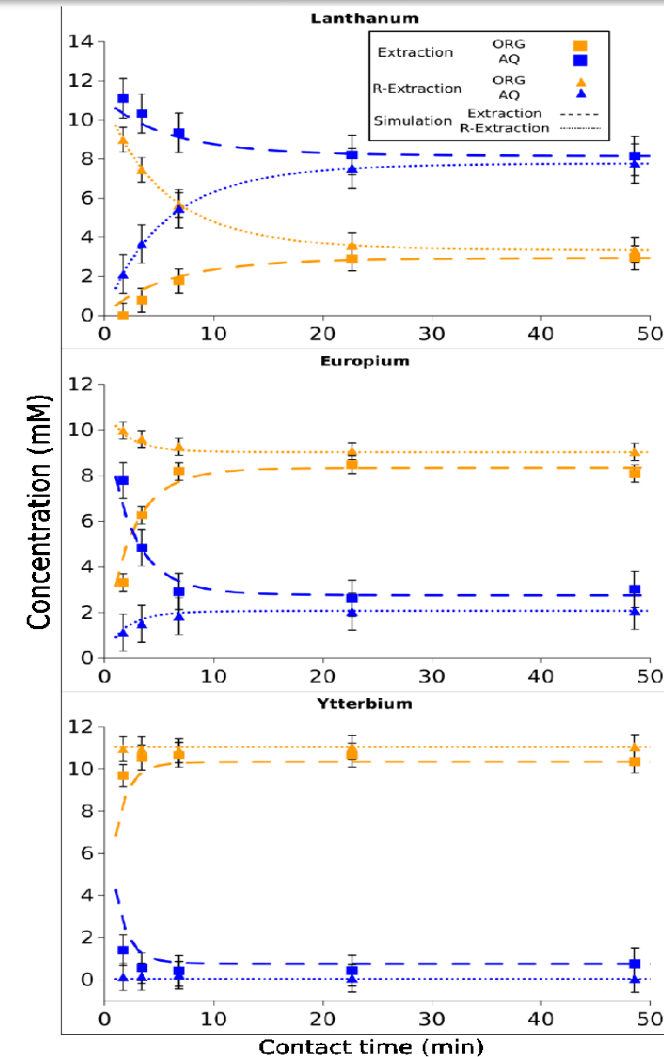
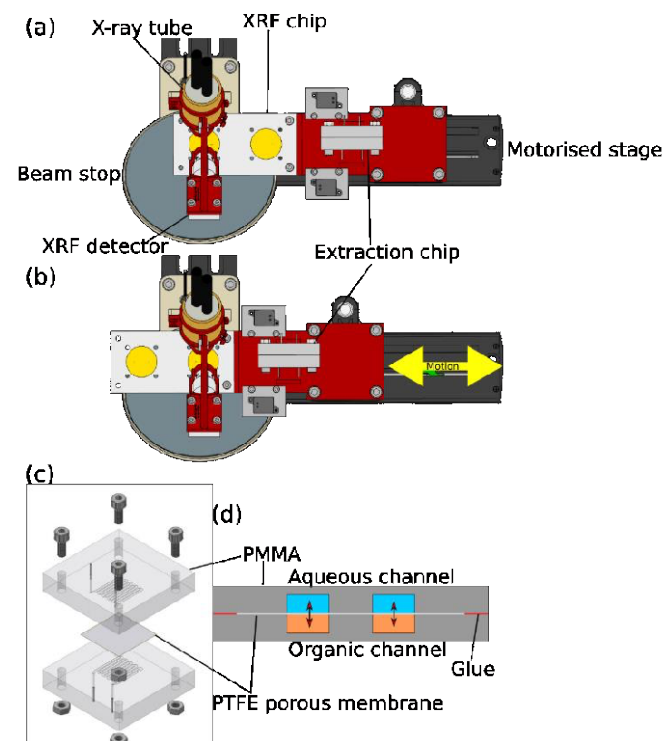
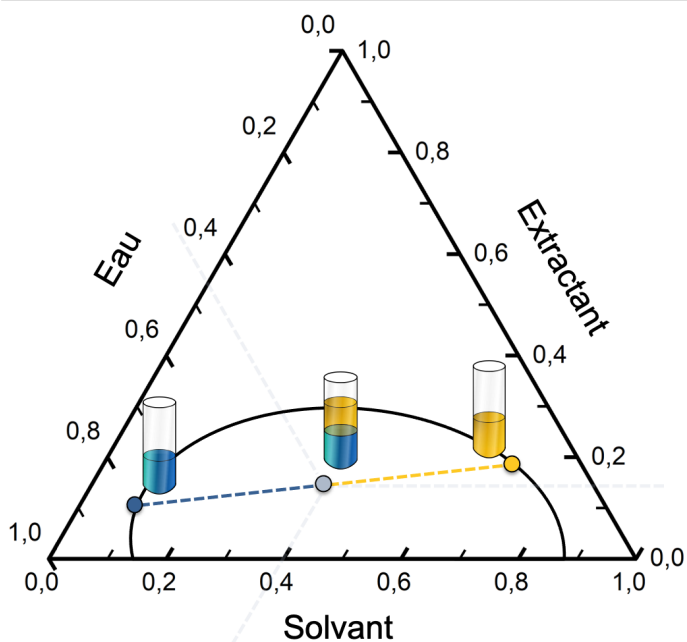


C. Breil, Th Zemnb, W. kunz & F. Chemat : **A green “Bligh and Dyer” and Folch Method for Solid–Liquid–Liquid Extraction of Lipids from Microorganisms: Comprehension of Solvation Mechanisms and towards Substitution with Alternative Solvents** *Int. J. Mol. Sci.* 2017



Replace « yield » % : $\Delta G_{tr} = RT \ln D = RT \ln ([w] / [o])$

First: necessity of XRF to have error bars of $\ln D < 0.5$ ($k_B T$)

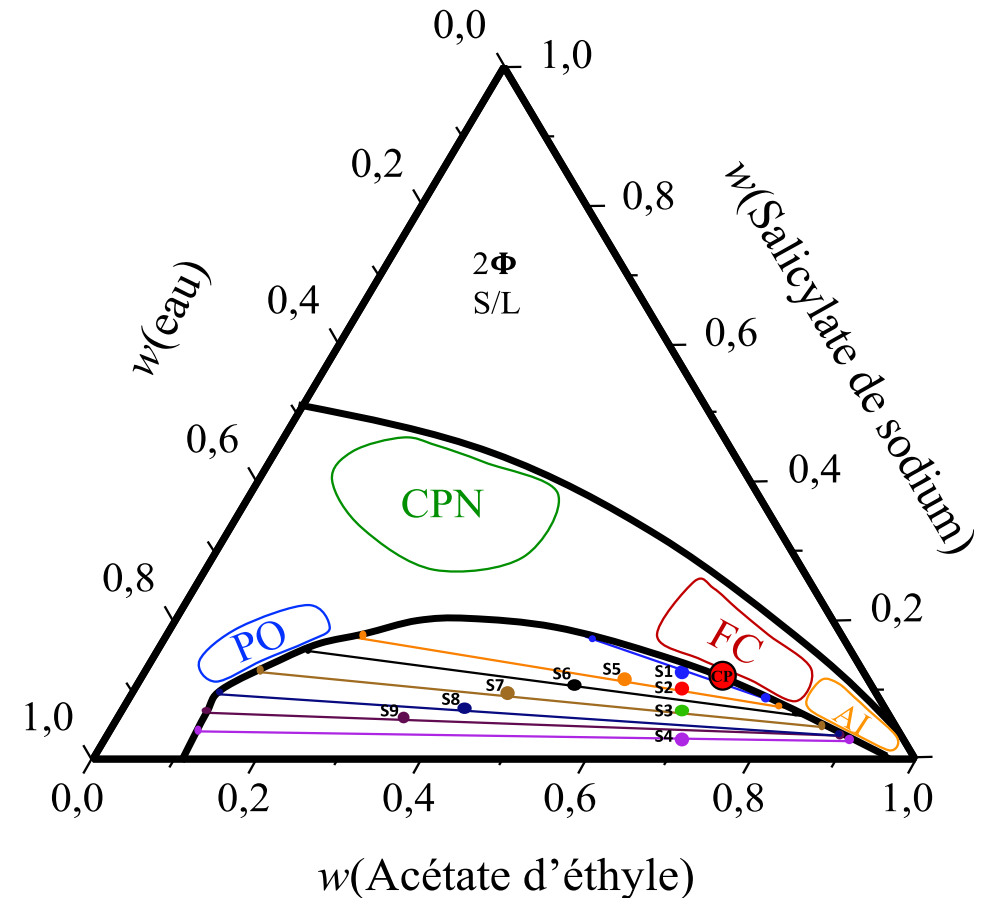


El Maangar, A.; Theisen, J.; Penisson, C.; Zemb, T.; Gabriel, J.-C. P.: **A Microfluidic Study of Synergic Liquid-Liquid Extraction of Rare Earth Elements**. PCCP 2020, 22 (10), 5449–5462. doi.org/10.1039/c9cp06569e.

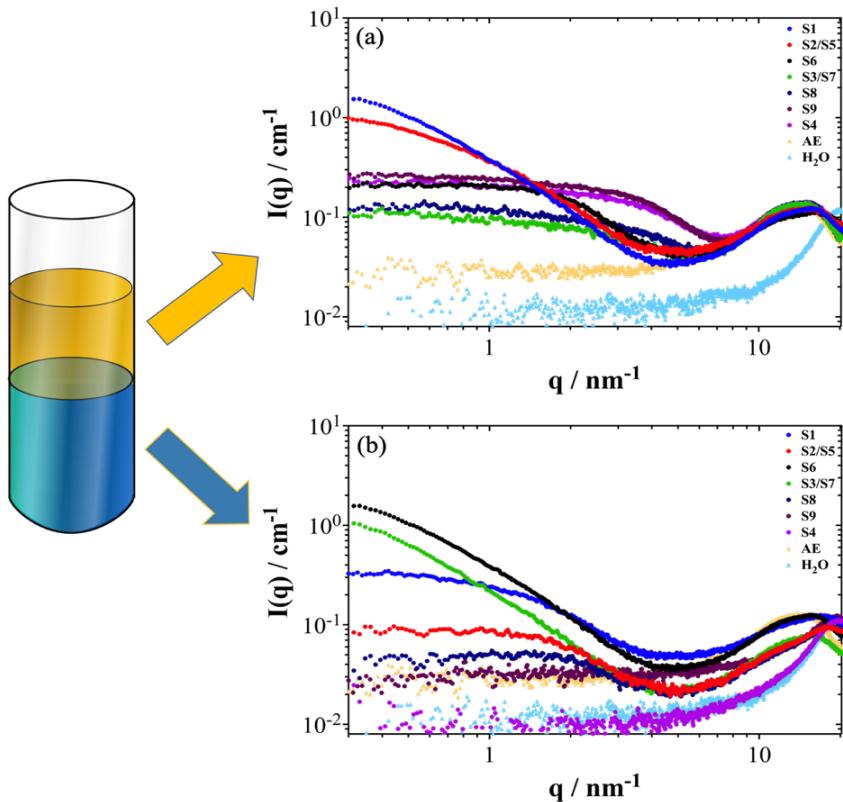
Precision in $\ln D$ using the “ienaic” approach



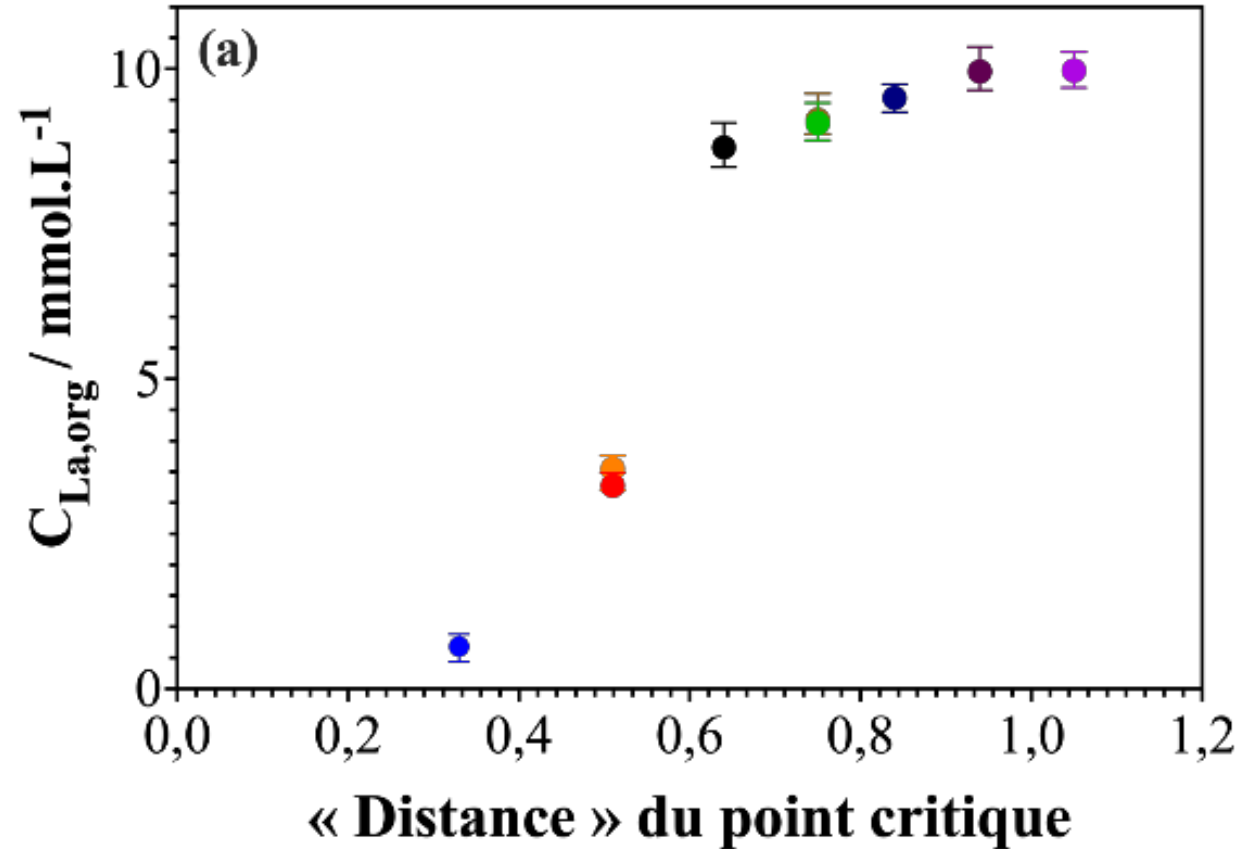
°97%+/-2 % « : means Incertitude in $RT\ln$ means D between 0.01 and 0.05 this means $\ln D$ is between 3 and 4.6 while target is determination of ΔG_{ext} to less than $0.5k_bT$ = XRF of both phases is the only known technique to achieve the precision required !



What is the yield in extraction: Yield **decreases** with **increase** of extractant in the solvent phase !!!!



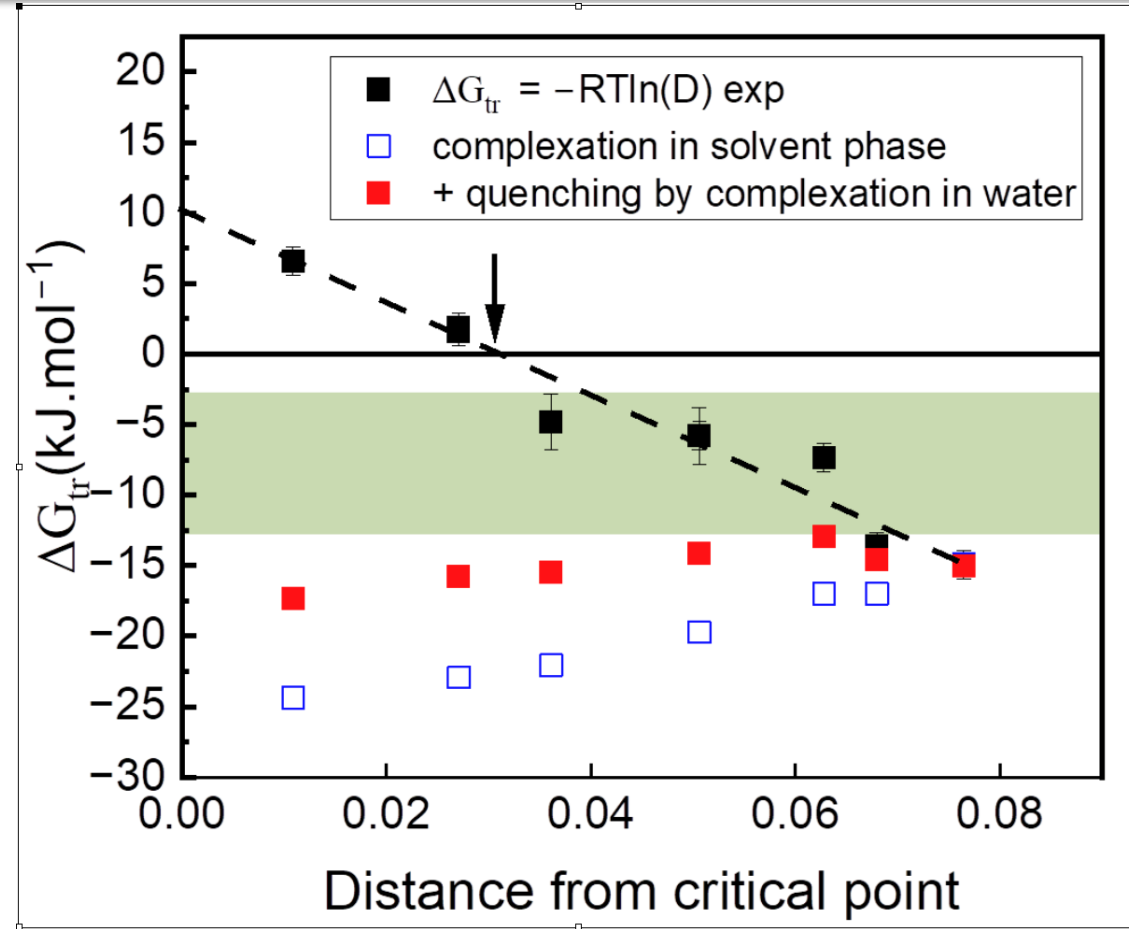
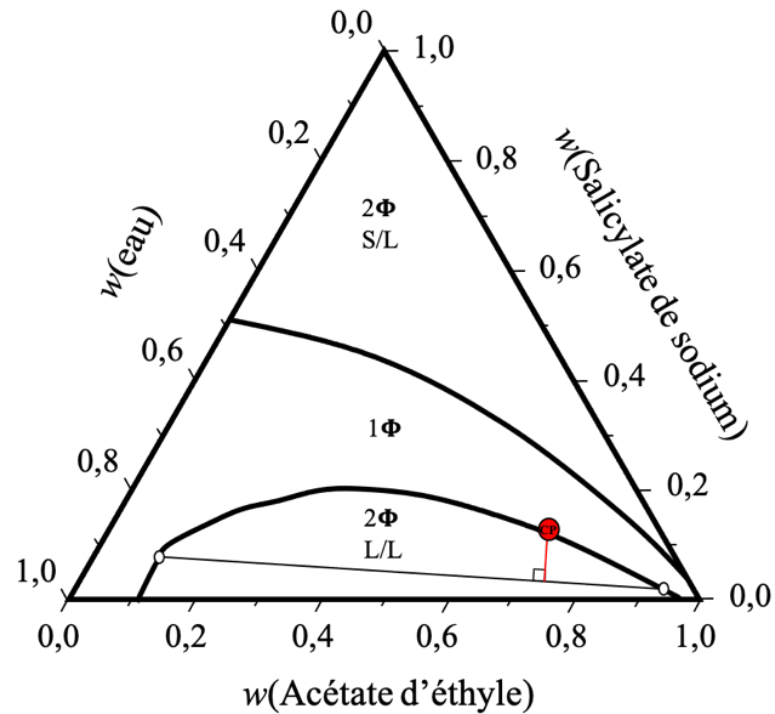
Identification of aggregation : i.e; self- assembly
(Or not) in coexisting phases..



- **El Maangar, A.**; Degot, P.; Huber, V.; Causse, J.; Berthault, P.; Touraud, D.; Kunz, W.; Zemb, T., *Pre-nucleation cluster formation upon ethyl acetate addition to an aqueous solution of an anionic hydrotrope*, *Journal of molecular liquids*, 2020,

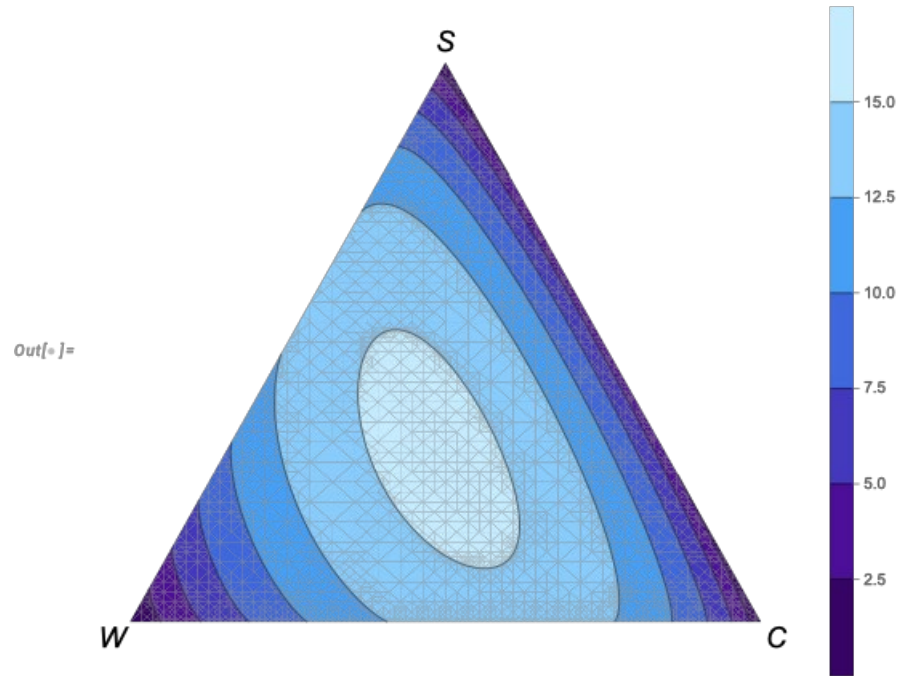
310 : 113240, DOI : 10.1016/j.molliq.2020.113240.

Ienaics approach: complexation plus solvent structure @ $T\Delta S$

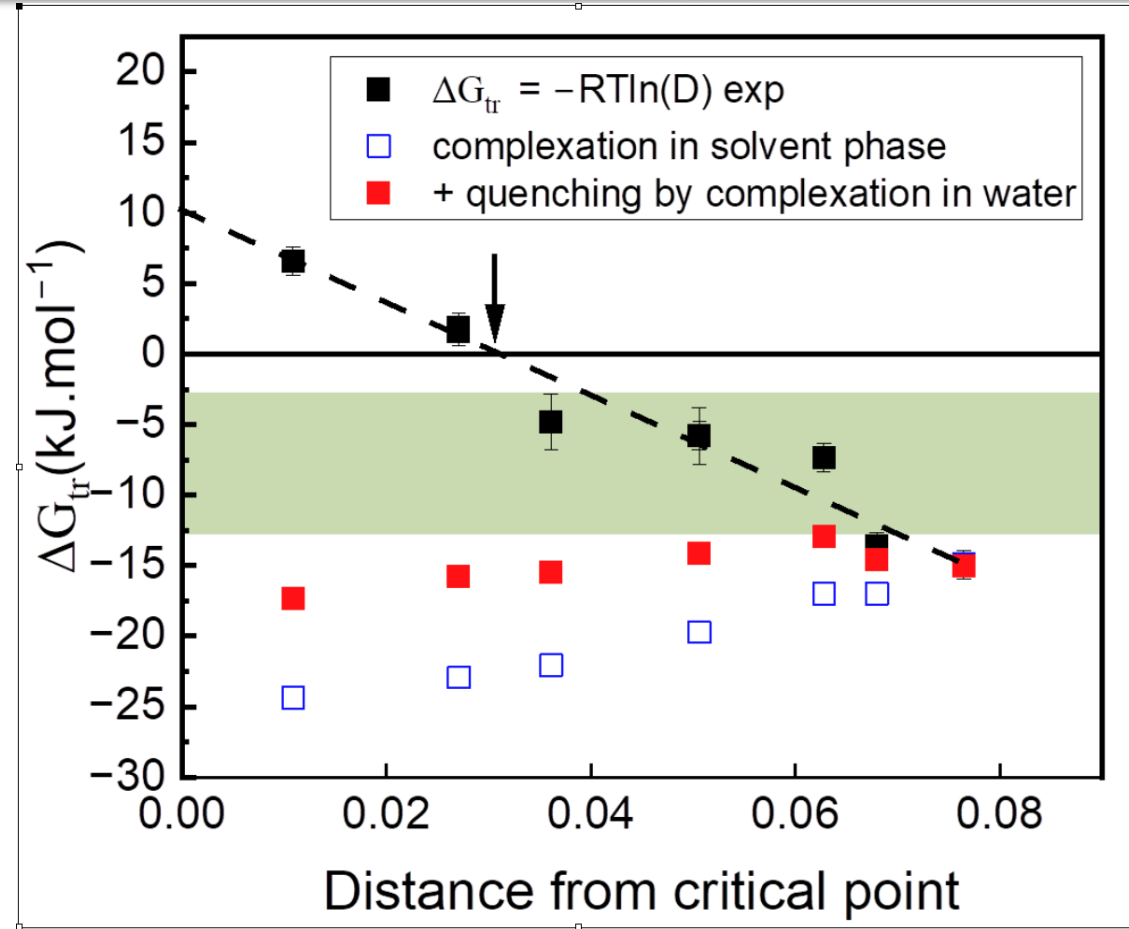


$$\Delta_{\text{tr}}G^{\circ} = \Delta_{\text{tr}}G^{\circ}_{\text{complex}} + \Delta_{\text{tr}}G^{\circ}_{\text{confin}} + \Delta_{\text{tr}}G^{\circ}_{\text{apolar}} + \Delta_{\text{tr}}G^{\circ}_{\text{lagg}} - T\Delta_{\text{tr}}S^{\circ}_{\text{pdisp}}$$

Ienaics approach: complexation plus solvent structure @ $T\Delta S$



S. Marcelja (2017) Phys. Rev.
Extensive intrinsic entropy of **ternary** fluids?
 $300 \times 0.5 \text{ J/nm}^3 \Rightarrow 5\text{-}10 \text{ kJ/mol}$.
Compare to 40-60 60KJ/mole complexation



$$\Delta_{\text{tr}}G^{\circ} = \Delta_{\text{tr}}G^{\circ}_{\text{complex}} + \Delta_{\text{tr}}G^{\circ}_{\text{confin}} + \Delta_{\text{tr}}G^{\circ}_{\text{apolar}} + \Delta_{\text{tr}}G^{\circ}_{\text{lagg}} - T\Delta_{\text{tr}}S^{\circ}_{\text{pdisp}}$$

A new technology for liq-liq extraction using hydrotopes as diluent modifier & co-complexant emerges

